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# Hydrogen Storage through Nanostructured Polymeric Materials

Di-Jia Liu<sup>1</sup>, Scott Kirklin<sup>1</sup>, Brian Dorney<sup>1</sup>, Suhas Niyogi<sup>1</sup>, Junbing Yang<sup>1</sup>, Peter Chupas<sup>1</sup>, Peter Zapol<sup>1</sup>, Shengwen Yuan<sup>2</sup> & Luping Yu<sup>2</sup>

> <sup>1</sup>Argonne National Laboratory <sup>2</sup>The University of Chicago

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Project ID ST21

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#### **Overview**



#### **Timeline**

Project start: July 2007Project end: June 2012

% complete: 20%

### **Budget**

- Total project funding: \$2 Million
  - DOE share: \$1.88 Million
  - Contractor share: \$120 K
- Funding received in FY07
  - \$ 200 K
- Funding for FY08
  - \$616 K

#### **Barriers**

- Barriers addressed
  - A. System Weight and Volume
  - B. System cost
  - C. Efficiency
  - D. Durability/Operability

#### **Partners**

- Interactions/collaborations
  - U of Chicago
  - DOE HSCoE Members
  - NREL
  - N. Carolina U.
  - Others



### **Objective**



- To design, synthesize, and evaluate nanostructured polymeric materials (NPM) as new hydrogen storage adsorbents for transportation applications
- To support polymer materials development with modeling/simulation and advanced structural characterizations

### Potential Advantages of Polymeric H2 Adsorbent

- Polymer surface properties such as specific surface area and porosity can be controlled at molecular level
- Polymer-hydrogen can be enhanced through incorporating different functional groups and atomically dispersed metals
- Polymers are generally stable under the temperature and humidity required for hydrogen storage application
- Polymer can be scaled-up for production with existing industrial infrastructure

#### Selected Literature Reports on Polymeric H<sub>2</sub> Adsorbent

- Cho et. al. Catal. Today, **2007** polyaniline and polypyrrole
- McKeown, et. al. Angew. Chem. Int. Ed. 2006 polymers with intrinsic porosity (PIMs)
- Wood, et. al. Chem Mater. 2007 hypercrosslinked polymers



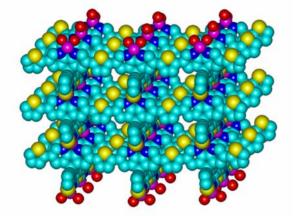


### **Approach**

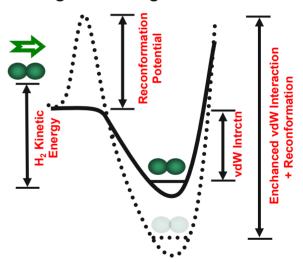


### Argonne/U of Chicago Approach to Polymer Design

- We will produce high surface area & narrow pore size through stereo-contorted polymer design
- We will incorporate "metallic" feature to polymer through conductive backbone
- We will improve polymer-H<sub>2</sub> interaction with by Introducing coordinated metal elements and various functional groups through synthetic approach
- We will explore the feasibility of "trapping" hydrogen through semi-rigid framework



An example of ANL/UofC polymer with conductive backbone incorporated with different elements



Control surface property and the interaction with hydrogen through design & synthesis at the molecular level!





### **Approach**



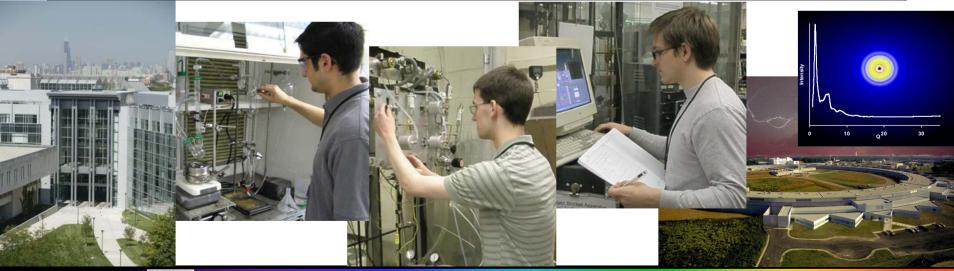
### New Polymer Exploration (UofC)

### Characterization & Optimization (ANL)

### Modeling & Simulation (ANL)

- New polymer synthesis through rational design at molecular level
- Molecular structure characterization
- H<sub>2</sub> storage capacity measurement
- Surface structure characterization
- Synthesis method improvement

- H<sub>2</sub>-polymer interaction study via *ab initio*, DFTB & MD methods
- Advanced X-ray characterization







### **Progress Summary**



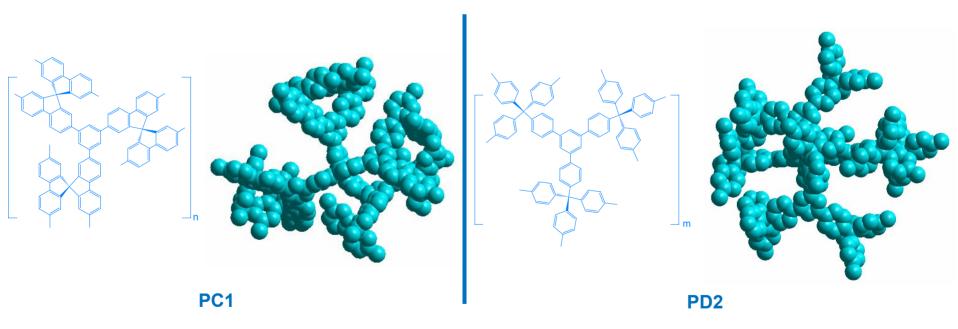
- Three series of porous polymers based on our design principles were successfully synthesized. Two groups showed promising surface properties (U of Chicago)
- Specific surface area and porosity of 11 polymer samples were measured, some of which showed high SSA and narrow pore distribution (Argonne)
- Upgrade of Sievert isotherm apparatus and analysis method were completed. System is now used for high pressure (up to ~75 bar) measurement (Argonne)
- Hydrogen uptake of several polymers were measured at both RT and 77K with pressure up to 75 bars. A carbon molecular sieve (CMS) material was also investigated as the reference material (Argonne)
- Theoretical modeling/simulation study was initiated (Argonne)



## Progress – Design & Synthesis of Porous Conjugated Polymers



- Seven porous polymers with conjugated aromatic planes were designed and synthesized
- Porosity was created and controlled through cross-linking of monomers with different stereo-contorted structures
- Polymer samples have demonstrated good surface properties and high stability towards heat and moisture



**Proposed Structures of the Selected Examples** 



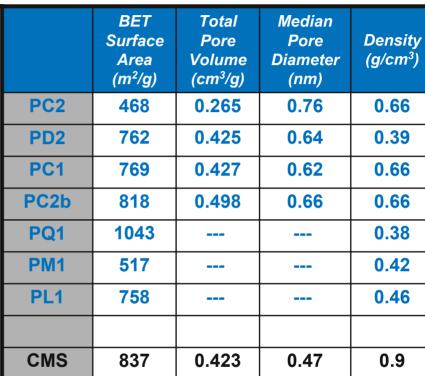


## Progress – Surface Property Characterization for Conjugated Porous Polymers

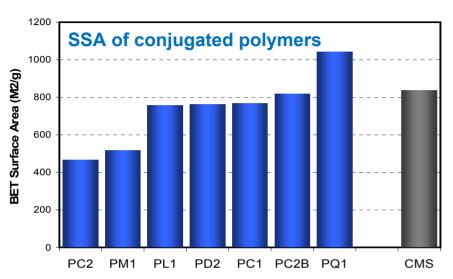


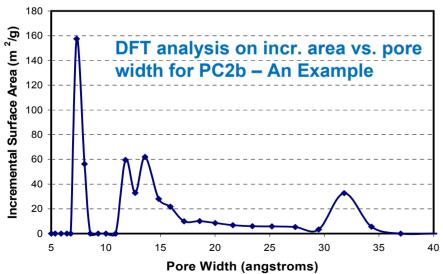






Relatively high surface area and narrow pore distribution can be achieved through rational polymer design.
Further improvement is still necessary.





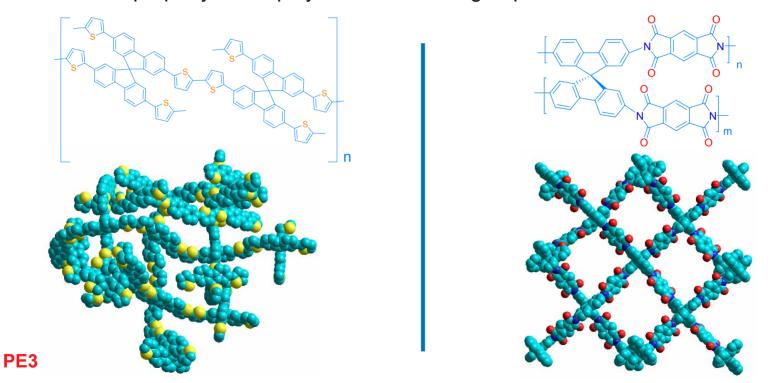




## Progress – Design & Synthesis of Polymers with Conductive & Polyimide Backbones



- Five porous polymers with conductive or polyimide backbones were designed and synthesized with the goal of enhancing hydrogen-functional group interaction
- Polymers prepared with spiral core linked by thiophene and bi-thiophene groups demonstrated relatively high surface area and promising pore size distribution
- The surface property of the polymers with imide group were mixed



**Proposed Structures of Selected Polymers** 



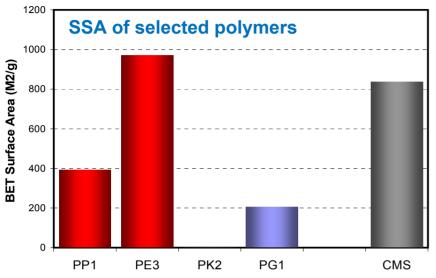


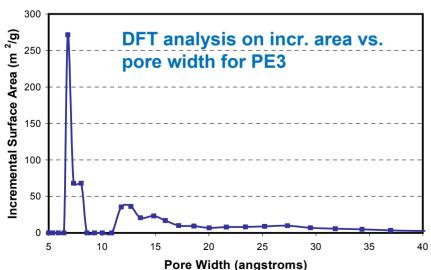
PG1

## Progress – Surface Property Characterization for Polymers Crosslinked with Conductive & Polyimide Backbones









#### **N<sub>2</sub>-BET Surface Characterization**

	BET Surface Area (m²/g)	Total Pore Volume (cm³/g)	Median Pore Diameter (nm)	Density (g/cm³)
PE3	971	0.575	0.62	0.30
PP1	393			0.19
PK2	2.1			0.55
PG1	206			0.42
CMS	837	0.423	0.47	0.9

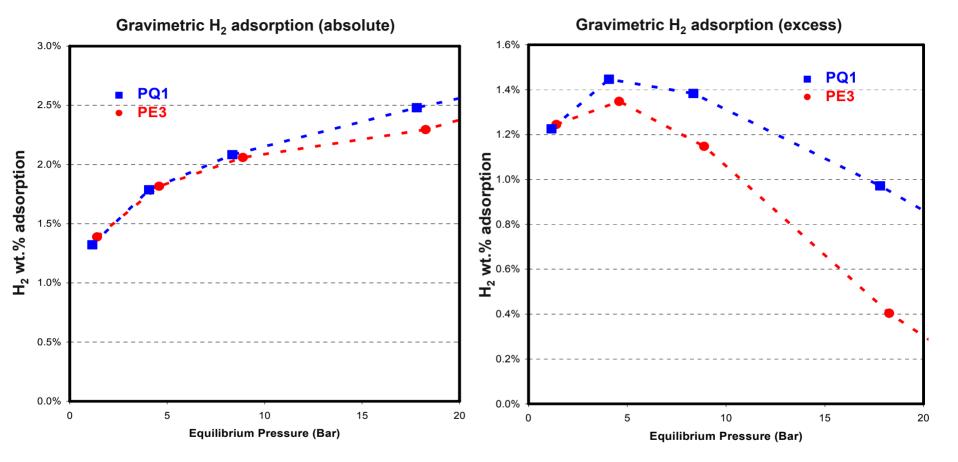
The synthesis condition has strong influence on the polymer structures during the condensation reaction





## Progress – H<sub>2</sub> Uptake Capacity Measurement at Liquid Nitrogen Temperature





Hydrogen uptake over the polymer surface at 77 K generally follows

Langmuir isotherm





## Progress – H<sub>2</sub> Uptake Capacity Measurement at Liquid Nitrogen Temperature



Sample	Gr. Uptake (Absolute)* (kg H <sub>2</sub> /kg adsorbent+H <sub>2ads</sub> )	Vol. Uptake (Absolute) (kg H <sub>2</sub> /L adsorbent)	Gr. Uptake (Excess) (kg H <sub>2</sub> /kg adsorbent+H <sub>2ads</sub> )	Vol. Uptake (Excess) (kg H <sub>2</sub> /L adsorbent)	BET SSA (M²/g)	Modified Chahine Factor
PQ1	2.1%	0.0080	1.4%	0.0053	1043	1.00
PC1	1.7%	0.011	1.2%	0.0079	769	1.11
PD2	1.2%	0.0047	0.5%	0.0020	762	0.77
PE3	2.1%	0.0063	1.2%	0.0036	971	1.06

<sup>\*</sup> Values in the table represent the measurement taken at a hydrogen pressure of ~ 7 bar

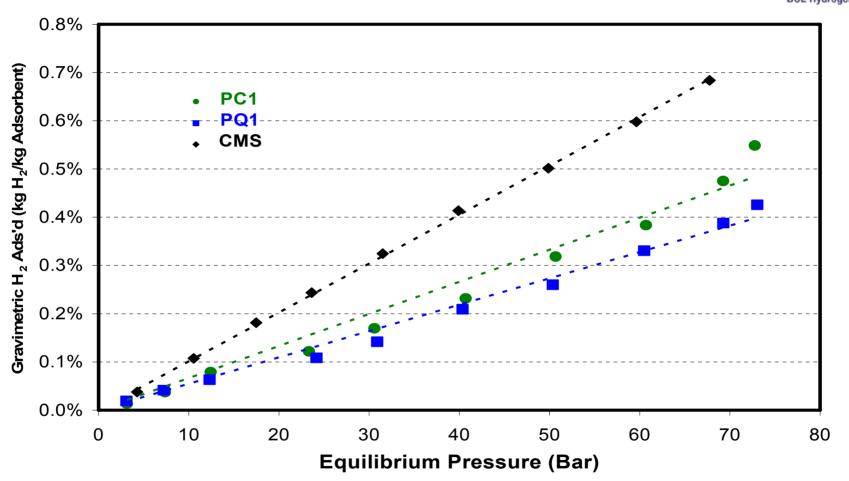
- Absolute capacities are generally proportional to surface area at 77 K.
- Excess capacities diminish when the storage pressure exceeds 20 bar; significant improvement in adsorption uptake is necessary.





## Progress – H<sub>2</sub> Uptake Capacity Measurement at Ambient Temperature





In contrast to CMS, H<sub>2</sub> uptakes of polymers deviate from linear increase as the function of pressure, suggesting a different H<sub>2</sub>/adsorbent interaction.





## Progress – H<sub>2</sub> Uptake Capacity Measurement at Ambient Temperature



Sample	Sample Mass Tested (gram)	H <sub>2</sub> Gravimetric Uptake @ ~70 Bar (kg H <sub>2</sub> /kg adsorbent+H <sub>2ads</sub> )	H <sub>2</sub> Volumetric Uptake @ ~70 Bar (kg H <sub>2</sub> /kg adsorbent+H <sub>2ads</sub> )
PQ1	0.360	0.43%	0.0016
PC1	0.220	0.55%	0.0036
PD2	0.192	0.66%	0.0026
PE3	0.239	0.45%	0.0014
CMS	0.430	0.54%	0.0049

- No clear correlation between H<sub>2</sub> uptake and surface area is observed
- Major improvements in capacities are necessary to reach the targets





### Progress – Investigation of Polymers from Literature



- Three polymer samples were prepared according to literature reports
- High yield was achieved through our modified synthesis process
- Surface characterization and hydrogen storage capacity measurement are underway

Potential mechanism for polymerization based on report by Wood, et. al. Chem. Mater. 2007, 19, 2034-2048





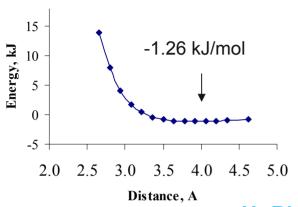
<sup>\*</sup> Potential active sites for alkylation. Activity changes after one of the sites within the aromatic group reacts

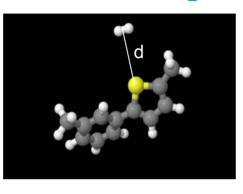
## Progress – Computational Modeling of Hydrogen Storage in Polymers



DOE Hydrogen Program

#### Calculated Structures & H<sub>2</sub> Binding for PE3

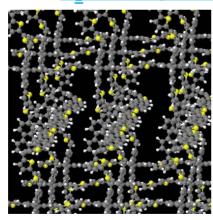




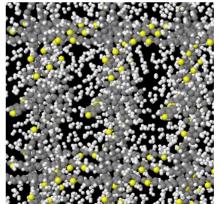
- Structures are optimized at B3LYP and MP2/631G(d) levels
- Hydrogen interaction energies are calculated at MP2/6311+G(2df,p) level

#### H<sub>2</sub> Binding inside of 3D Polymer Structure

3D cross-linked structure optimized with DFT-PW91







Ab initio MD at high H content

- Local H<sub>2</sub> binding energies over open surface of functional group are low
- Simulation on H<sub>2</sub> distribution, polymer deformation, etc., in 3D space is planned





#### **Milestones**



- Complete 1st phase experimental optimization study on TBHTP system 09/07 √
- Complete the initial design and synthesis of one new polymeric system 09/07  $\sqrt{\phantom{a}}$
- Provide one or more polymer samples to a DOE laboratory outside of Argonne for hydrogen storage capacity measurement
   01/08 √
- Complete the surface property characterization and hydrogen storage capacity measurement of the benchmark materials to compare with Argonne/U of Chicago polymers 06/08 →
- Complete first PDF experiments trial on polymer/hydrogen interaction at the Advanced Photon Source 07/08 →
- Initiate theoretical simulation of the interactions between hydrogen and model polymer systems 08/08 →
- Complete design and synthesize two or more new porous polymer materials with targeted hydrogen uptake capacity of 3% at 77 K and 1% at 298 K 08/08 →

#### **Future Work**



#### **FY08**

- Complete storage measurement and surface property characterization for all the new porous conjugated polymers prepared by U of C team. Derive a preliminary understanding of the molecular structure-capacity relationship
- Complete the storage capacity and structural characterization of the polymers prepared according to the literature report.
- Continue to improve the measurement accuracy of Sievert apparatus for study at higher pressure region (up to 100 bar)
- Initiate preliminary X-ray PDF experiment to probe polymer structure and its interaction with hydrogen under elevated pressure

#### **FY09**

- Continue polymer design and synthesis by preparing two additional systems of a) polymers with different main group elements and b) polymers with metal doping
- Optimize synthesis conditions of the best systems to further enhance the surface property and storage capacity.
- Continue computational modeling of hydrogen-polymer interaction and the conformation change at the elevated pressure
- Collaborate with other members of HSCoE in adsorption mechanistic studies (e.g. NMR, neutron, etc.)





### **Summary**



Relevance:

Developing the nanostructured porous polymers as new H<sub>2</sub> storage media aimed at meeting DOE performance targets for transportation applications.

Approach:

Rational design and synthesis at molecular level supported by surface property/storage capacity measurement, computational modeling and advanced characterization.

**Accomplishments**:

- Three series of polymeric adsorbents designed and prepared with high surface area and narrow pore distribution.
- H<sub>2</sub> uptakes up to 75 bar studied at both 77K and RT; interesting adsorption behavior observed.

Collaboration:

Core team of Argonne and U of Chicago collaborating with HSCoE members.

**Future Work:** 

- Continuous polymer exploration and optimization.
- Improving capacity measurement & mechanistic study at high H<sub>2</sub> pressure.
- Understanding possible conformation change through theory and advanced characterization.

